Enzymatic Preparation of Specifically Modified Linear Maltooligosaccharides through Taka-amylase A-catalyzed Hydrolysis of 6-O-Arenesulfonyl- \$7\$-cyclodextrins

Kahee FUJITA, \* Tsutomu TAHARA, \* Taiji IMOTO, \* \* and Toshitaka KOGA \* Faculty of Pharmaceutical Sciences, Fukuyama University,

Sanzo, Higashimura-cho, Fukuyama 729-02

\*Daiichi College of Pharmaceutical Sciences,

Tamagawa, Minami-ku, Fukuoka 815

\*\*Faculty of Pharmaceutical Sciences, Kyushu University,

Maidashi, Higashi-ku, Fukuoka 812

 $6-O-Arenesulfonyl-\gamma-cyclodextrins$  were hydrolyzed enzymatically by Taka-amylase A to give 6"-O-arenesulfonylated maltotetraoses and 6"-O-arenesulfonylated maltotrioses.

Cyclodextrins are enzymatically hydrolyzed by some  $\alpha$ -amylases including Taka-amylase A.<sup>1)</sup> Melton and Slessor reported that 6-0-substituted  $\alpha$ -cyclodextrins were hydrolyzed by Taka-amylase A to give selectively 6'-0-substituted maltoses.<sup>2)</sup> We also reported that 6-0-substituted  $\alpha$ - or  $\beta$ -cyclodextrins,<sup>3)</sup> 2-0-substituted  $\alpha$ - or  $\beta$ -cyclodextrins,<sup>4)</sup> 3-0-substituted  $\alpha$ - or  $\beta$ -cyclodextrins,<sup>4)</sup> or  $3^A$ ,  $6^A$ -anhydro- $\beta$ -cyclodextrin<sup>5)</sup> gave 6'-0-substituted maltoses, 2"-0-substituted maltotrioses, 3'-0-substituted maltotrioses (and/or 3"-0-substituted maltotetraose), and 3",6"-anhydromaltotetraose, respectively. These studies provided new information about interactions between substrates and the active sites of Taka-amylase A as well as a novel synthetic method for the preparation of specifically modified (or activated) oligosaccharides.

However, these experiments using  $\alpha$ - or  $\beta$ -cyclodextrin derivatives give only the final products and therefore very limited information about the subsites of the active site. The reason is that the slow rate of hydrolytic cleavage of the cyclodextrin macrocyclic ring does not allow accumulation of intermediate linear oligosaccharides in enough amount to be isolated.

Since  $\gamma$ -cyclodextrin is hydrolyzed much faster than  $\alpha$ - or  $\beta$ -cyclodextrin, enzymatic hydrolysis of  $\gamma$ -cyclodextrin derivatives by large amount (not catalytic amount) of Taka-amylase A can reasonably be expected to give new specifically modified linear oligosaccharides (new intermediates) which can not be isolated in the enzymatic hydrolysis of modified  $\alpha$ - or  $\beta$ -cyclodextrins. This result will also provide information concerning the importance of the subsites.

Table 1 shows the condition and result of the Taka-amylase A-catalyzed hydrolysis of 6-0-arenesulfonyl- $\gamma$ -cyclodextrin 1a-c which was easily prepared by the reaction of  $\gamma$ -cyclodextrin with the corresponding arenesulfonyl chloride in pyridine.  $^{6)}$ 

1a:  $X = \beta - Ns0$ 

 $1b: X = \alpha - Ns0$ 

1c: X = p-Ts0

Ns = naphthalenesulfony1

Ts = toluenesulfonyl

2a: 
$$m = 1$$
,  $n = 1$ ,  $X = \beta-NsO$ 

2b:  $m = 1$ ,  $n = 1$ ,  $X = \alpha-NsO$ 

2c:  $m = 1$ ,  $n = 1$ ,  $X = \beta-NsO$ 

3a:  $m = 0$ ,  $n = 1$ ,  $X = \beta-NsO$ 

3b:  $m = 0$ ,  $n = 1$ ,  $X = \alpha-NsO$ 

3c:  $m = 0$ ,  $n = 1$ ,  $X = \beta-NsO$ 

4a:  $m = 0$ ,  $n = 1$ ,  $X = \beta-NsO$ 

4b:  $m = 0$ ,  $n = 0$ ,  $X = \beta-NsO$ 

4c:  $m = 0$ ,  $n = 0$ ,  $X = \beta-NsO$ 

5:  $m = 0$ ,  $n = 0$ ,  $X = \beta-NsO$ 

The products 2a-c, 3a-c, and 4a-c were isolated by reverse-phase column chromatography. Their FAB mass spectra showed the corresponding molecular ions. By treatment with aqueous alkali, the products 2a-c were converted to 3",6"-anhydromaltotetraose whose structure was determined by comparing its HPLC retention time and its FAB mass spectrum with those of the authentic compound. The products 3a-c were reduced at their reducing ends with NaBH<sub>4</sub> followed by complete acetylation with acetic anhydride in pyridine. From analysis of the EI mass spectra of the resultant undecaacetates, the position of the sulfonylated glucose unit in 3a-c was determined. The products 4a-c were converted to 6'-S-phenyl-6'-thiomaltose 5 whose structure was determined by comparing the HPLC retention time and the FAB mass spectrum with those of the authentic specimen. 3)

Chemistry Letters, 1988

Cyclodextrin derivative	Temp/ <sup>O</sup> C	Time	Product(Yield/%)b)
<b>1</b> a	10	2 h	<b>2</b> a (67.4) <sup>c)</sup>
	10	2 d	<b>2</b> a (28.8), <b>3</b> a (28.9)
	40	3 h	<b>4</b> a (88.3)
<b>1</b> b	10	1.5 h	2b (35.5), <b>3</b> b (54.4)
	40	4 h	<b>4</b> b (61.1)
<b>1</b> c	10	2 h	2c (31.7), 3c (43.6)
	10	2 d	<b>4</b> c (99.4)

Table 1. Enzymatic Hydrolysis of 6-O-Substituted  $\Upsilon$ -Cyclodextrin by Taka-amylase  $A^{a}$ )

a) The solvent was 0.1 mol dm $^{-3}$  acetate buffer (pH 5.5) containing CaCl $_2$  (0.01 mol dm $^{-3}$ ). Taka-amylase A( $\alpha$ -amylase from Aspergillus oryzae) which is commercially available as  $\alpha$ -amylase Type X-A (Sigma) was used. The concentrations of the cyclodextrin derivative and Taka-amylase A were each 10 mg/(1 mL of the solvent). b) This represents the major product(s) having the arenesulfonyl group. Needless to say, D-glucose was another major product. c) The progress of the reaction was stopped just before 3a began to form.

The specifically activated (sulfonylated) oligosaccharides 2a-c and 3a-c could not be isolated from the enzymatic hydrolysis of 6-O-arenesulfonyl- $\alpha$ - or  $\beta$ -cyclodextrins, even when Taka-amylase A was used in an amount equal to that of the cyclodextrin derivative 2,3 and/or the reaction was carried out at 5 °C or 10 °C. On the other hand, purely chemical methods for the preparation of these compounds from smaller sugars will need troublesome repeats of activation, protection, and deprotection of hydroxyl groups of the sugars. One-step chemical synthesis of specifically modified linear oligosaccharides from the corresponding linear oligosaccharides is also impossible. This method will require tedious separation of many regioisomeric monosulfonylated oligosaccharides.

Since there does not exist any regiochemical isomer, 6-O-arenesulfonyl- $\gamma$ -cyclodextrins can be easily prepared and easily isolated as a pure material. As shown above, they can be smoothly hydrolyzed to give one or two modified oligosaccharide(s) at 10 °C. Furthermore, the products can be effectively separated by reverse-phase column chromatography. Therefore, the selective chemical modification of  $\gamma$ -cyclodextrin followed by the Taka-amylase A-catalyzed hydrolysis has several advantages for preparation of specifically modified oligosaccharides.

Although the reaction of 1a at 10  $^{O}$ C for 2 h afforded only the 6"-0-substituted maltotetraose (2a), the hydrolysis at 10  $^{O}$ C for longer reaction time (2 d) or at 40  $^{O}$ C gave a mixture of 2a and 3a or exclusively 4a, respectively.

1332 Chemistry Letters, 1988

The hydrolysis of 1b at 10  $^{\circ}$ C for 1.5 h gave a mixture of 2b and 3b. The reaction of 1c at 10  $^{\circ}$ C for 2 h afforded a mixture of 2c and 3c. However, the hydrolysis of 1b or 1c at higher reaction temperature and/or for longer reaction time gave 4b or 4c, respectively. These show that the 6"-O-substituted maltotetraose (2a-c) and the 6"-O-substituted maltotriose (3a-c) were the intermediates and that the 6"-O-substituted maltotetraose was, at least in the reaction of 1a, the precursor of the 6"-O-substituted maltotriose, which led to the 6'-O-substituted maltose.

These results will be useful for substrate-based investigation on interaction between hydroxyl groups of substrates and the enzyme's active site which consists of seven subsites.<sup>8)</sup> This will be reported in the near future together with other results.

We are indebted to Japan Maize Products Co. Ltd. for a generous gift of  $\gamma$ -cyclodextrin.

## References

- 1) K. Suetsugu, S. Koyama, K. Takeo, and T. Kuge, J. Biochem., 76, 57 (1974).
- 2) L. D. Melton and K. N. Slessor, Can. J. Chem., <u>51</u>, 327 (1973).
- 3) K. Fujita, A. Matsunaga, and T. Imoto, Tetrahedron Lett., 1984, 5533; K. Fujita, A. Matsunaga, Y. Ikeda, and T. Imoto, ibid., 1985, 6439; K. Fujita, A. Matsunaga, and T. Imoto, J. Am. Chem. Soc., 106, 5740 (1984).
- 4) K. Fujita, T. Tahara, S. Nagamura, T. Imoto, and T. Koga, J. Org. Chem., <u>52</u>, 636 (1987).
- 5) K. Fujita, H. Yamamura, T. Imoto, and I. Tabushi, Chem. Lett., 1988, 543.
- 6) For example, the tosylate (1c) was prepared as follows. p-Toluenesulfonyl chloride (12 g) was added to a solution of γ-cyclodextrin (10 g) in pyridine (800 mL) and the mixture was stirred for 1 h. After addition of water (10 mL), the mixture was concentrated in vacuo and chromatographed by a reverse-phase column (LiChroprep RP18 column, 25 mm x 310 mm, Merck) with gradient elution from water (1 L) to 50% aqueous ethanol (1 L) to give the monotosylate (3.5 g, 31%). The other arenesulfonates 1a,b were similarly prepared and isolated in similar yields.
- 7) A Merck prepacked column (see Ref. 6) or a Kusano prepacked ODS column (CPO-HS-221-20, 22 mm x 100 mm) was used for reverse-phase column chromatography. After elution with water to eliminate inorganic salts and glucose, gradient elution from water to aqueous methanol or aqueous acetonitrile was used for separation of the products having the arenesulfonyl group.
- 8) K. Hiromi, Biochem. Biophys. Res. Commun.,  $\underline{41}$ , 1 (1970); J. A. Thoma, C. Brothers, and J. E. Spradlin, Biochemistry,  $\underline{9}$ , 1768 (1970).

(Received May 16, 1988)